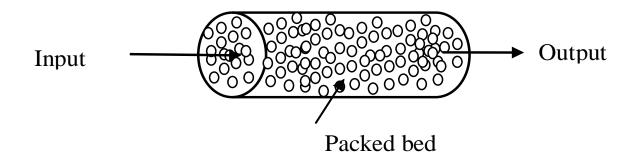
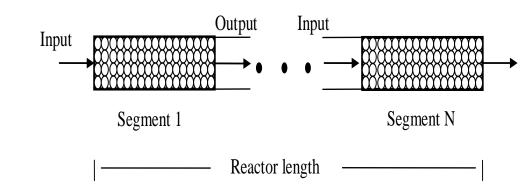
# Multiphysics Simulation of a Packed Bed Reactor

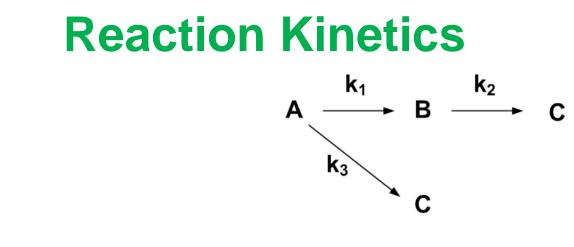
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### **Schematic of a Packed Bed Reactor**



### **Segmented geometry**





A = o-xylene, B = phthalic anhydride, C = carbon monoxide and carbon dioxide  $k_i$ , *i*=1, 2, 3, = rate constants

$$r_{1} = \rho_{b} y_{o} y_{Ao} k_{1} (1 - x_{A})$$
  

$$r_{2} = \rho_{b} y_{o} y_{Ao} k_{2} x_{B}$$
  

$$r_{3} = \rho_{b} y_{o} y_{Ao} k_{3} (1 - x_{A})$$

Kinetics

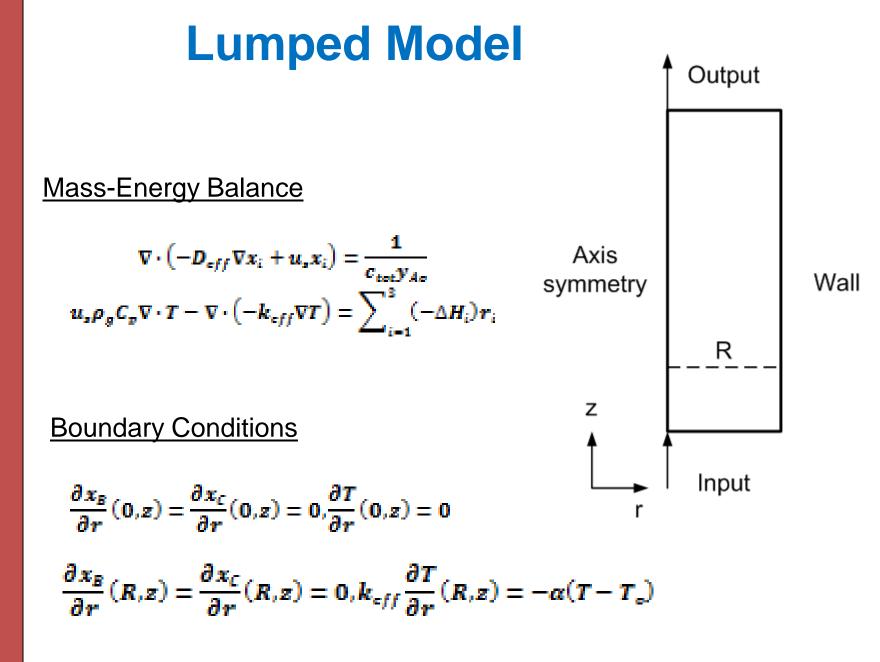
### **Kinetics Parametres**

(Lerou and Froment,, Chem. Eng. Science, vol. 32, 1977)

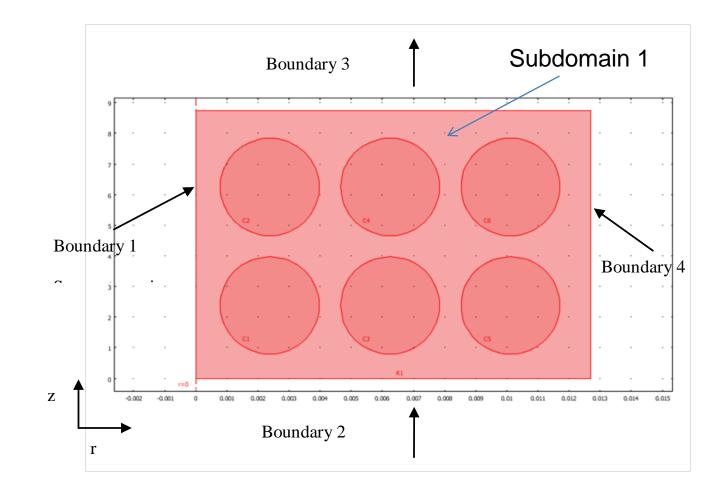
$$k_i = A_i \exp\left(-\frac{B_i}{T}\right), \quad (i = 1, 2, 3)$$

 $A_1 = \exp(19.837) \text{ kmol/kg h}$   $A_2 = \exp(20.86) \text{ kmol/kg h}$   $A_3 = \exp(18.97) \text{ kmol/kg h}$   $B_1 = 13588 \text{ K}$   $B_2 = 15803 \text{ K}$  $B_3 = 14394 \text{ K}$ 

 $\Delta H_1 = -1.285 \times 10^6 \text{ J/mol}$   $\Delta H_2 = -3.276 \times 10^6 \text{ J/mol}$  $\Delta H_3 = \Delta H_1 + \Delta H_2$ 



### **Segment Model**



Length = 0.00873 m, radius = 0.0127 m, circle radius = 0.00159 m, uniform separation = 0.00079 m

### Navier-Stokes equations - Subdomain 1

 $\rho_{g} u \cdot \nabla u = \nabla \cdot \left[ -p\mathbf{I} + \mu(\nabla u + (\nabla u)^{T}) \right]$   $\nabla \cdot u = 0$ Initial conditions  $\begin{cases} u_{to} = 0 \\ v_{to} = 0 \\ p_{to} = p_{o} \end{cases}$ 

#### **Convection and diffusion equations**

Subdomain 1

 $\nabla \cdot (-D\nabla c_B) = -u \cdot \nabla c_B, u = [u \ v]^7$  $\nabla \cdot (-D\nabla c_C) = -u \cdot \nabla c_O u = [u \ v]^7$ 

Subdomains 2, .., 7 (circles)

$$\nabla \cdot \left(-D_{eff} \nabla c_B\right) = r_B / y_{Ao'}$$
$$\nabla \cdot \left(-D_{eff} \nabla c_C\right) = r_C / y_{Ao}$$

Initial conditions  $c_{Bto} = 0, c_{Cto} = 0$ 

#### **Energy balance equations**

Subdomain 1

$$\nabla \cdot (-k \cdot \nabla T) = -\rho_g C_{vg} u \cdot \nabla T;$$
$$u = [u \ v]$$

Subdomains 2, ..., 7 (circles)  $\nabla \cdot (-k_{eff} \cdot \nabla T) = (-\Delta H_1)r_s + (-\Delta H_2)r_c - \rho_s C_s u \cdot \nabla T$ Initial condition  $T_{to} = T_0$ 

#### **BCs / Navier-Stokes equations** $p_{to} = p_o$ Outlet $\mu(\nabla u + (\nabla u)^T)n = 0$ Boundary 3 Subdomain 1 Symmetry No slip Boundary 1 u = 0Boundary 4 $n \cdot u = 0^{\circ}$ $t \cdot [-p\mathbf{I} + \mu(\nabla u + (\nabla u)^{T})]n = 0$ Ζ -0.002 -0.001 0.011 0.012 0.013 0.014 0.015 Boundary 2 r $u_o = 0$

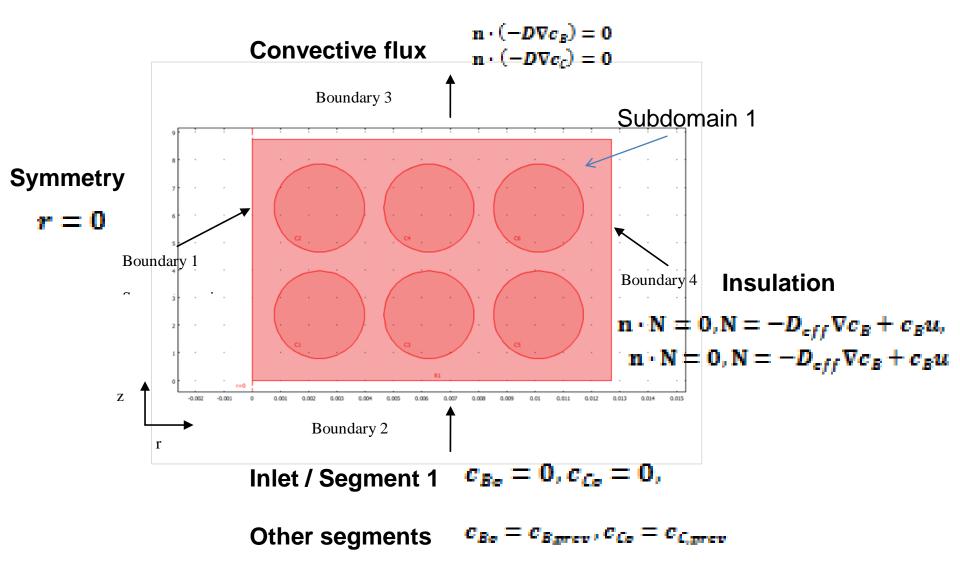
Other segments  $\begin{cases} u \\ v \end{cases}$ 

Inlet / Segment 1

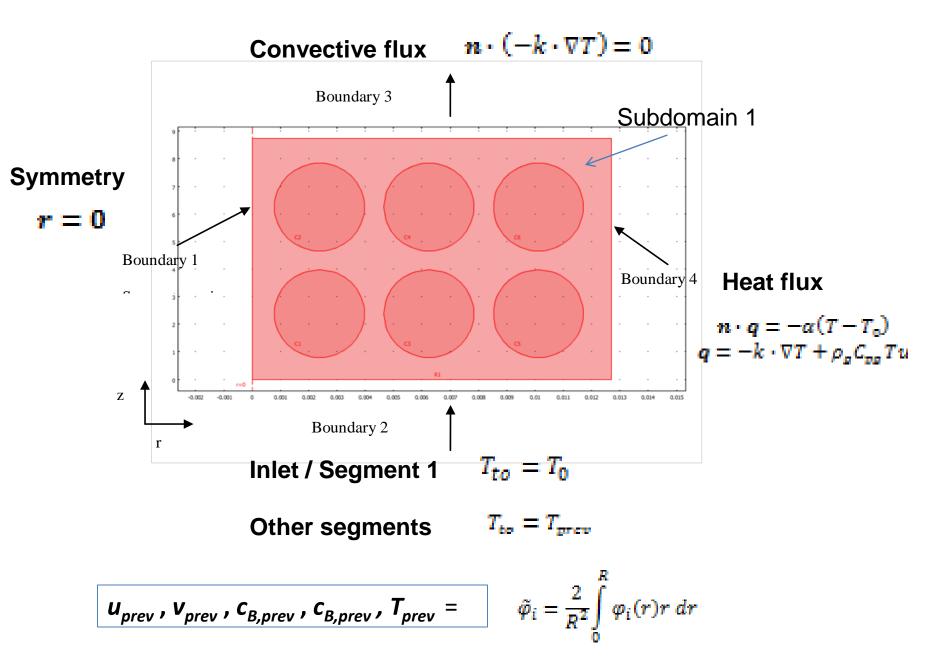
 $\begin{cases} u_o = u_{prev} \\ v_o = v_{prev} \end{cases}$ 

 $(v_n = u_n)$ 

### **BCs / Convection and diffusion equations**



### **BCs / Energy balance equations**



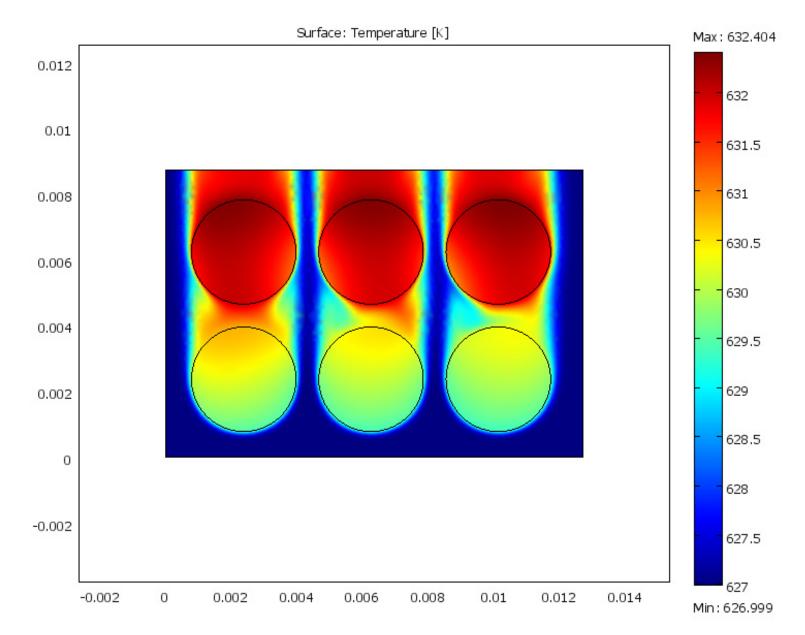
### Approach

- **a.** First set up the 2D-axis-symmtry geometry and application modes.
- **b.** Mesh generation by selecting a maximum element size of  $1 \times 10^{-4}$  m for boundaries 1, 2, 3 and 4, and  $0.5 \times 10^{-5}$  for interior circle boundaries.
- **c.** Next set up subdomains 1 and subdomains 2-6 for the different application modes and boundary conditions.
- **d.** Now solve the problem. First choose the Navier-Stokes equations to compute the velocity components (*u*, *v*) for subdomain 1. Then save variables and solve energy and mass balance equations.
- e. Save the above process as a matlab file and modify it for the successive changes in boundary and input conditions of each reactor segment.
- **f.** Execution of the matlab file and analysis of results under.

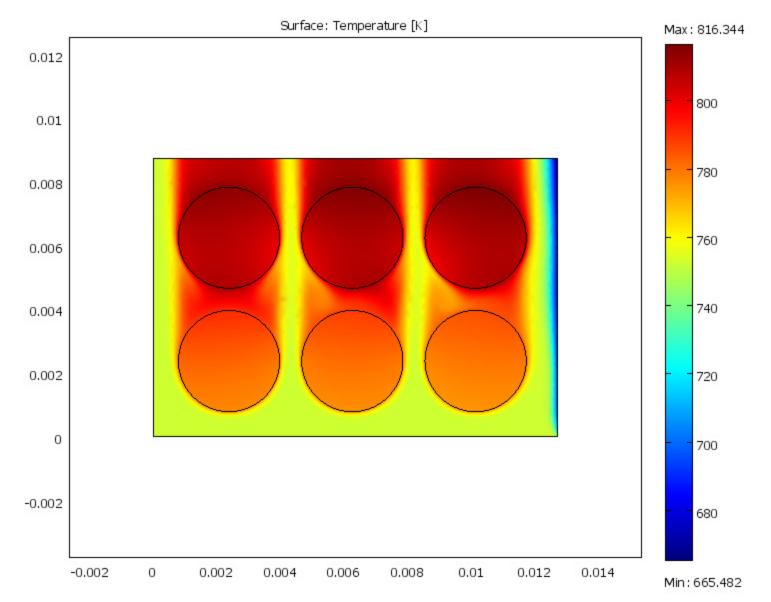
### **Model Parameters**

Reactor radius, R = 1.27 cm, Superficial velocity,  $v_s = 1.064$  m/s, (m/s), Inlet temperature,  $T_0 = 627$  K, Pressure,  $p_0 = 1.013 \times 10^5 \text{ Pa}$ Inlet total concentration,  $c_{tot} = 44.85 \text{ mol/m}^3 \text{ mol/m}$ , Inlet mole fraction of o-xylene,  $y_{Ao} = 0.00924$ , Inlet mole fraction of oxygen,  $y_0 = 0.208$ , Catalyst bulk density  $\rho_{b} = 1300 \text{ kg/m}^{3}$ , Gas density  $\rho_q = 1293 \text{ kg/m}^3$ , Gas heat capacity  $C_p = 1046 \text{ J/(kg K)}$ , Heat transfer coefficient  $\alpha = 156 \text{ W/(m}^2 \text{ K})$ . Effective diffusion constant,  $D_{eff} = 3.19e-7 \text{ m}^2/\text{s}$ , Diffusivity of xylene in air,  $D = 8.074 \times 10^{-5} \text{ m}^2/\text{s}$ , Effective thermal conductivity,  $k_{eff} = 0.779$  W/(m K), Gas thermal conductivity, k = 0.0318 W/(m K).

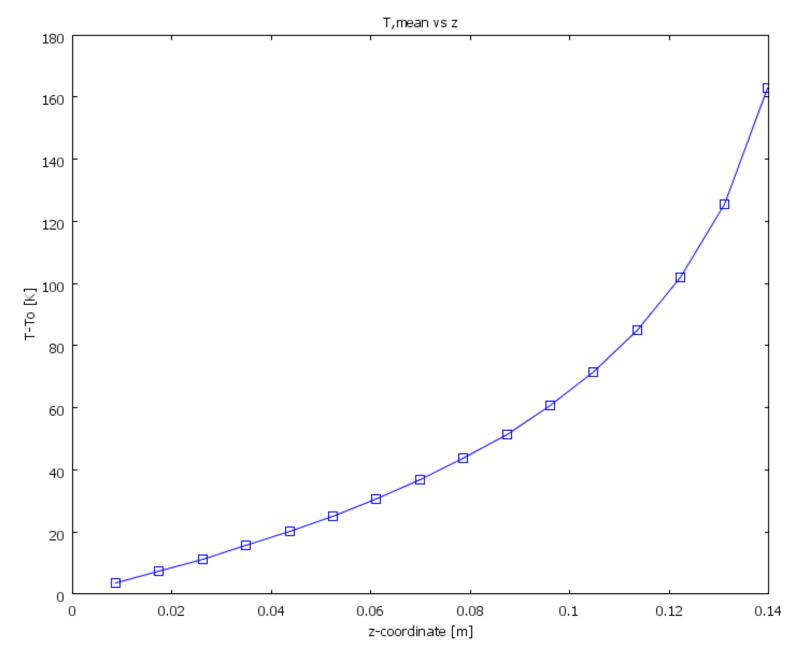
### Temperature distribution, segment 1, L = 0.0873 m



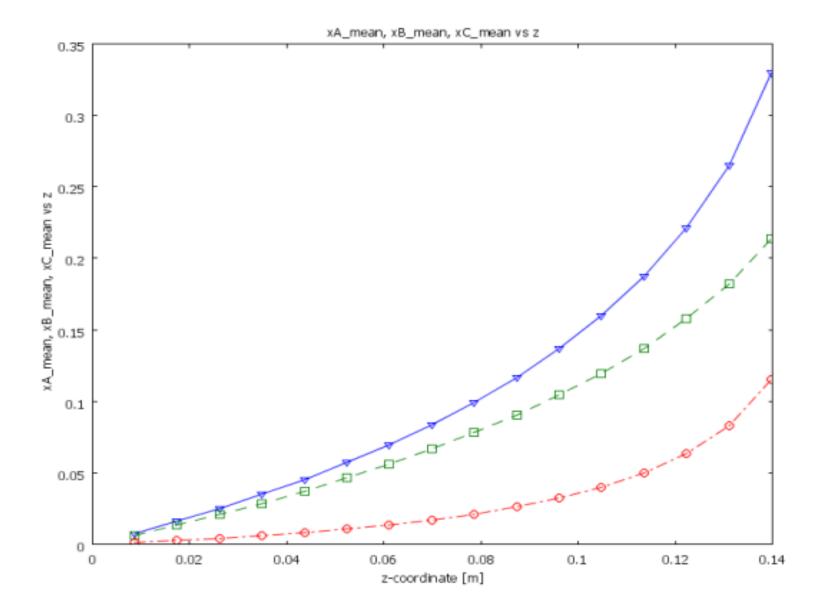
#### Temperature distribution, segment 16, L = L = 0.13968 m



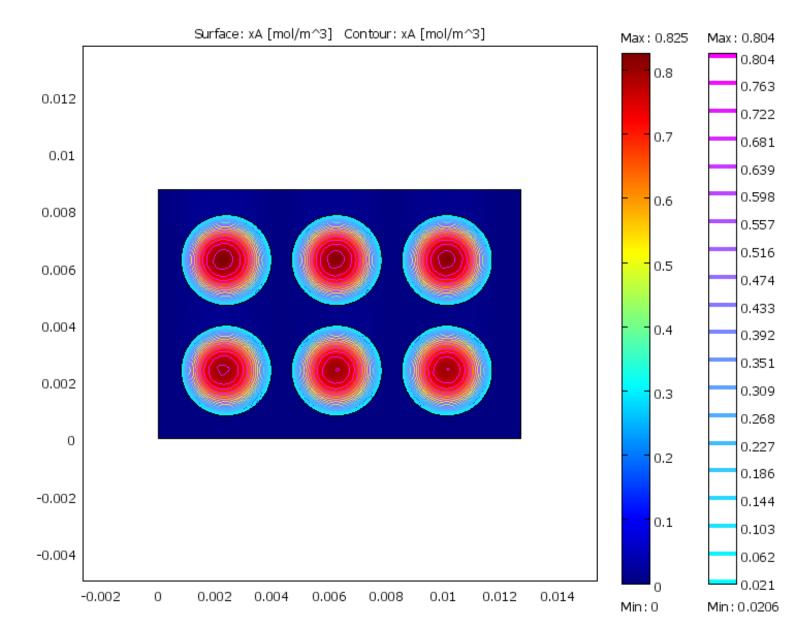
#### Average temperature profile, L = 0.13968 m



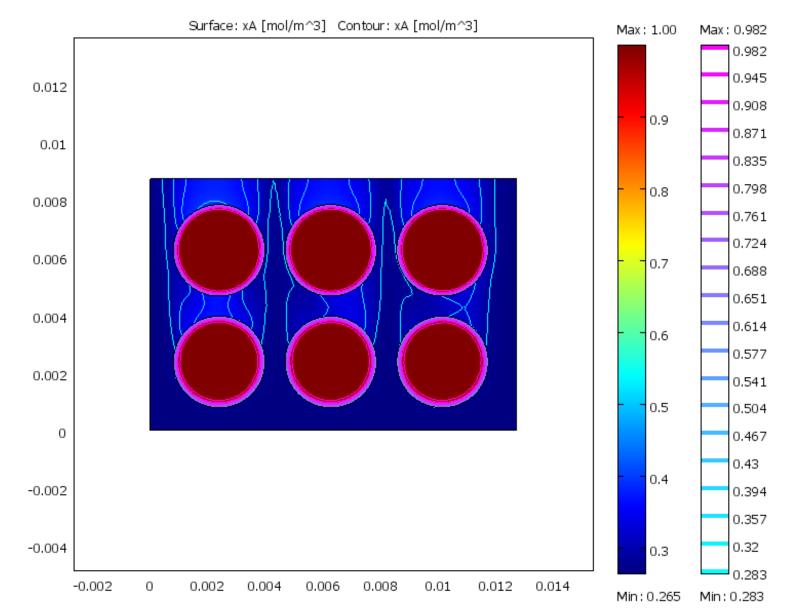
#### Average conversions, *L* = 0.13968 m



#### Conversion of A, segment 1, L = 0.0873 m



### Conversion of A, segment 16, L = 0.13968 m



### Conclusion

High temperatures increases or hot-spots occurred near reactor inlet. In order to study possible undesired conversions and catalyst damage, a more detailed model is desirable. For a future work we suggest consideration of a geometry model with intra pellet gaps and in contact with each other for the packed bed reactor simulation. This study should include sensitivity analysis of inlet temperature and heat transfer through the wall.

## **Many Thanks**